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1,1,1,3,3,3-Hexafluoro-2-(4-nitrophenyl)-*N*-phenylisopropylamine

Stephen F. Pavkovic,* Derek W. Nelson and James R. Owens

Department of Chemistry, Loyola University Chicago, 6525 N. Sheridan Road, Chicago, IL 60626, USA

Correspondence e-mail: spavko1@luc.edu

Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(\text{C-C}) = 0.003 \text{ Å}$ R factor = 0.059 wR factor = 0.173Data-to-parameter ratio = 25.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

There is no interaction between the amine H and any F atom in the title compound, $C_{15}H_{10}F_6N_2O_2$. Rather the amine H atom is hydrogen bonded to one of the nitro group O atoms on a neighboring molecule. This hydrogen bonding operates in the direction of the minor axis and links molecules into infinite chains.

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Comment

(Trifluoromethyl)trimethylsilane reacts with nitrones to afford α -(trifluoromethyl)hydroxylamines, protected as O-trimethylsilyl ethers. Adducts with strong electron-withdrawing groups on the α -aryl ring undergo an elimination/addition sequence to produce α,α -bis(trifluoromethyl)amines. The structure of compound, (I), produced from the p-nitrophenyl adduct, is reported here.

Experimental

The preparation of (I) has been reported (Nelson *et al.*, 2001). Crystals suitable for diffraction analysis were obtained from hexane.

Crystal data

 $C_{15}H_{10}F_6N_2O_2$ D_m : more dense than CCl₄ $M_r = 364.25$ D_m measured by flotation Monoclinic, C2/c Mo $K\alpha$ radiation a = 16.903 (3) Å Cell parameters from 13 b = 10.182 (2) Åreflections $\theta = 10-12.5^{\circ}$ c = 17.971 (3) Å $\mu = 0.16 \text{ mm}^{-1}$ $\beta = 104.90 (1)^{\circ}$ $V = 2989.1 (9) \text{ Å}^3$ T = 293 (2) KCut fragment, pale yellow Z = 8 $D_x = 1.619 \text{ Mg m}^{-3}$ $0.40 \times 0.35 \times 0.25 \text{ mm}$

Data collection

Picker four-circle diffractometer $\theta/2\theta$ scans $k=-7 \rightarrow 13$ $I=-23 \rightarrow 23$ 3603 independent reflections $I=-23 \rightarrow 23$ 3633 independent reflections 4 standard reflections 273 reflections with $I>2\sigma(I)$ every 150 reflections 28 intensity decay: <1% $\theta_{\rm max}=28.0^\circ$

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organic papers

Refinement

 $\begin{array}{lll} \mbox{Refinement on } F^2 & w = [\exp[1.30(\sin\theta/\lambda)^2]]/[\sigma^2(F_o^2) \\ R[F^2 > 2\sigma(F^2)] = 0.059 & + (0.0559P)^2] \\ wR(F^2) = 0.173 & where <math>P = 0.33F_o^2 + 0.67F_c^2 \\ S = 0.98 & (\Delta/\sigma)_{max} = 0.010 \\ 3603 \ \mbox{reflections} & \Delta\rho_{max} = 0.51 \ \mbox{e Å}^{-3} \\ 143 \ \mbox{parameters} & \Delta\rho_{min} = -0.26 \ \mbox{e Å}^{-3} \\ \mbox{H-atom parameters constrained} & Extinction correction: $SHELXL97$ \\ Extinction coefficient: 0.0018 (3) \\ \end{array}$

Table 1 Hydrogen-bonding geometry (Å, °).

| D $ H$ $\cdot \cdot \cdot A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdot \cdot \cdot A$ | D $ H$ $\cdot \cdot \cdot A$ |
|--------------------------------|------|-------------------------|-------------------------|--------------------------------|
| N1-H1···O1 ⁱ | 0.86 | 2.42 | 3.234 (4) | 159 |

Symmetry code: (i) x, y - 1, z.

Both phenyl rings were refined as rigid groups.

Data collection: *XSTAL* (Brown, 1985), a modification of Picker FACS–I software (Picker, 1967); cell refinement: *CELL* (Brown, 1985); data reduction: *MINCON* (Brown, 1985); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP*-3 (Farrugia, 1997).

References

Brown, J. N. (1985). Personal communication. Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.

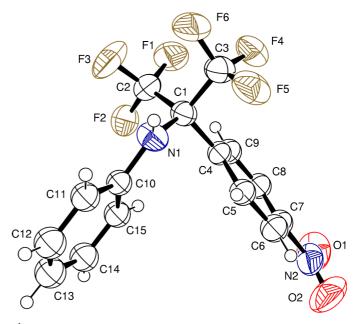


Figure 1 Displacement ellipsoid plot of the title compound with ellipsoids at the 50% probability level and H atoms omitted for clarity.

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